

Supporting Information

SYNTHESIS OF ONE COVALENT ORGANIC FRAMEWORK (COF) BASED ON C=N BONDS AND ITS EXCELLENT PERFORMANCE OF IODINE ADSORPTION

Yingbin Zhang, Limei Li, Meige Wang, and Zhongyu Duan*

School of Chemical Engineering, Hebei University of Technology, 1 Dingzigu
Road, Tianjin, 300130, China. E-mail: zyduan@hebut.edu.cn

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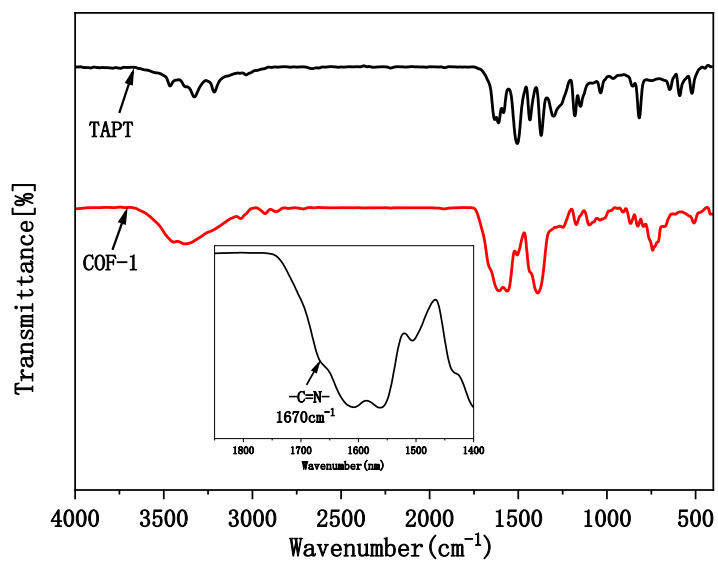


Figure S1 FT-IR spectra of TAPT and COF-1

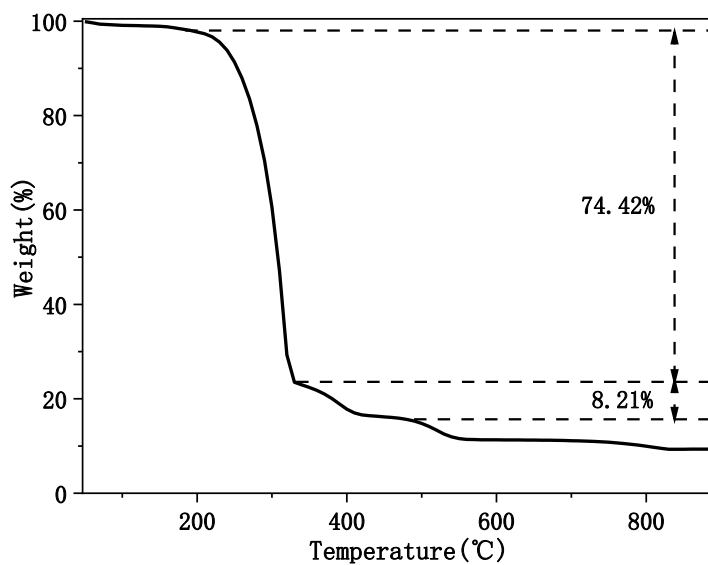
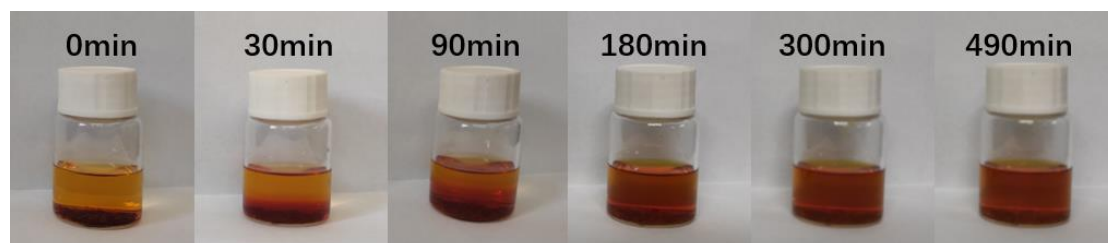


Figure S2 TG curves of COF-1

Table S1 Comparison of iodine uptake of different adsorbents

Adsorbents	Temperature(K)	Iodine uptake(mg/g)	Ref.
NTP	348	300	1
Ag ⁺ -MOR	368	275	2
Ag@Mon-POF	343	250	3
CMPN-2,-3	343	1100,2080	4
Azo-Trip	350	2330	5
Nip-CMP	350	2020	6
ZIF-8	350	1250	7
Cu-BTC	348	1750	8
CC3	293	364	9
PAF-21	348	1520	10
[Cu ₆ (pybz) ₈ (OH) ₂]	413	760	11
[Fe ₃ (HCOO) ₆](I ₂) _{0.84}	298	488	12
MoSx	333	1000	13
Polyurethane(PU1)	343	1284	14
3D-PPy	353	1600	15
PAF-1	298	1860	16

**Figure S3** I₂@COF-1 release in anhydrous methanol

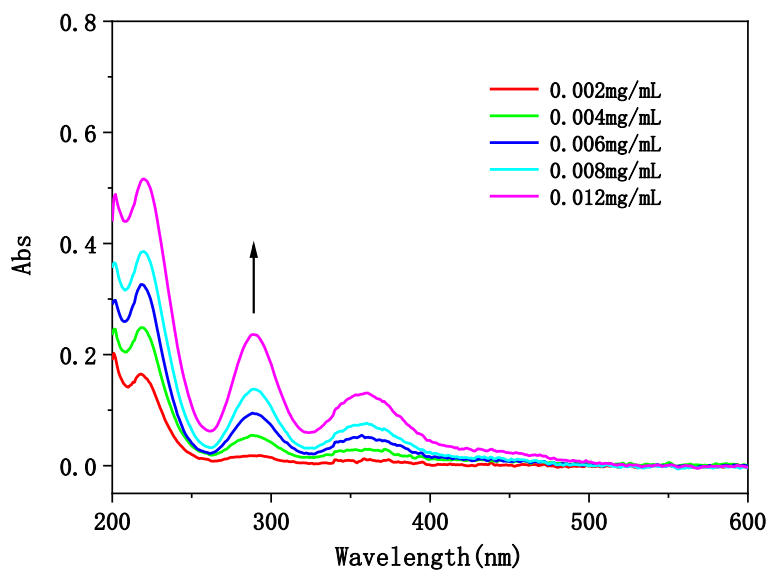


Figure S4 Calibration plot of standard I₂ by UV-vis spectra in methanol solution

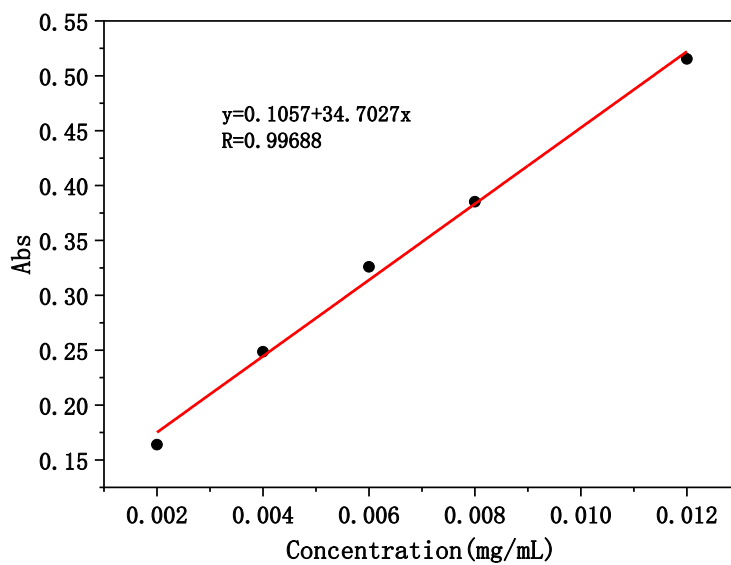


Figure S5 The fitting of Abs value vs concentration of I₂ with the relatively good linearity satisfies Lambert-Beer Law

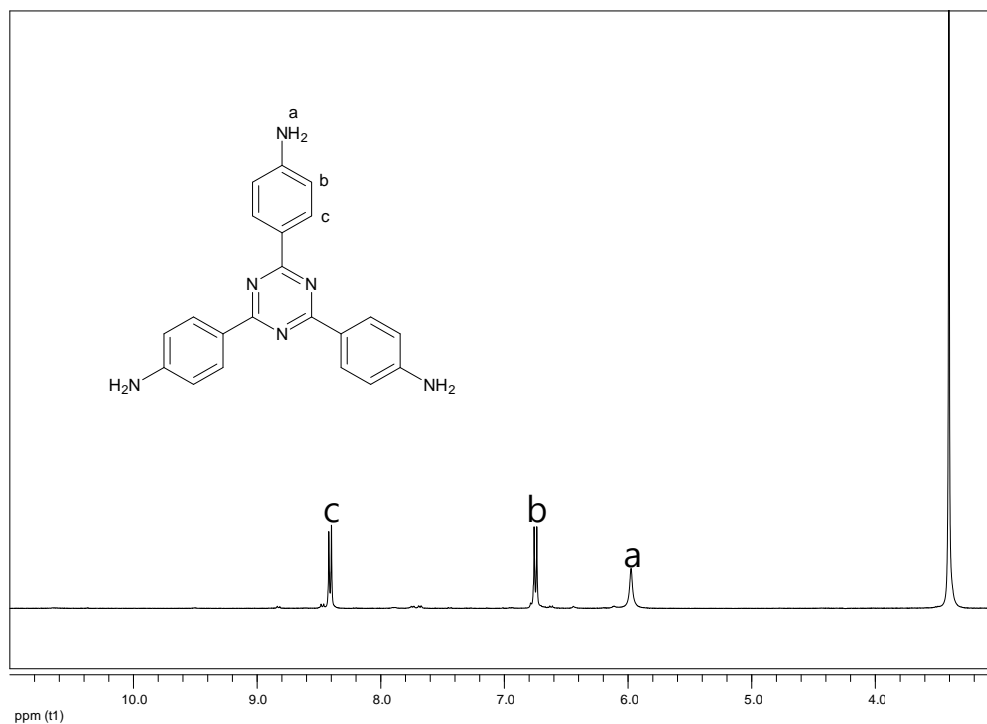


Figure S6 ^1H NMR of 1,3,5-tri-(4-amino-phenyl) triazine (TAPT)

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