

TWO NEW SPIROBENZYLISOQUINOLINE ALKALOIDS FROM RUPICAPNOS AFRICANA (LAM.) POMEL

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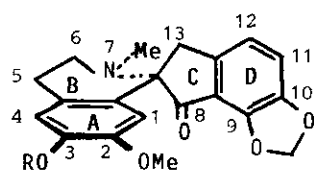
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Abstract- Two new spirobenzylisoquinolines, (+)-isoparfumine and (+)-africanine, have been isolated from Rupicapnos africana (Lam.) Pomel and identified spectroscopically.

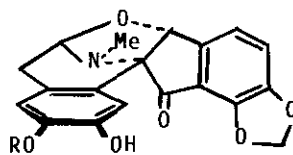
The spirobenzylisoquinolines are a group of alkaloids that have hitherto been isolated almost exclusively from the genera Fumaria and Corydalis¹. As part of our work on the isolation and identification of Fumarioideae alkaloids² we have recently studied specimens of Rupicapnos africana (Lam.) Pomel collected in the Sierra del Chorro (Málaga) which as well as the known isoquinolines protopine, aobamidine and coptisine³ were found to contain two new spirobenzylisoquinolines that we have named (+)-isoparfumine (1) and (+)-africanine (3).

(+)-Isoparfumine (1) was obtained as a white, crystalline, optically active substance, mp 206-208°C (MeOH), $[\alpha]_D^{25} +54^\circ$ (c= 0.792, CHCl₃). Its UV spectrum, with absorptions at λ_{\max} (EtOH) (log ϵ) 210(4.30), 234(4.35), 260(4.60), 293(sh) and 350(3.47) nm, λ_{\max} (EtOH+OH⁻) 300 and 350 nm and a band in its IR spectrum at λ_{\max} (CHCl₃) 1710 cm⁻¹ were typical of an 8-ketospirobenzylisoquinoline^{1,4}. The IR spectrum also displayed a band at 3500 cm⁻¹ (OH). The molecular formula C₂₀H₁₉O₅N, obtained by elemental analysis, was confirmed by MS, in which the molecular ion appeared at m/z 353 (52%). In addition, the fragment (M⁺-29) characteristic of an 8-ketospirobenzylisoquinoline⁵ was observed at 324 (100%). The pmr data (80 MHz, acetone-d₆, δ) suggested structure (1) for (+)-isoparfumine in view of the singlets at 2.28 (3H, NMe), 3.52 (3H, OMe), 6.15 (2H, OCH₂O), 6.26 (1H, H-1), 6.59 (1H, H-4) and an AB centred at 6.93 and 7.20 ppm (J= 7.9 Hz, H-12 and H-11 respectively). Further support for this assignment was obtained by O-methylation of (+)-isoparfumine with diazomethane, which afforded (+)-parfumidine (2)², whose identity with an authentic specimen was proved by direct comparison (IR, NMR, tlc). Finally, the methoxy group of (+)-isoparfumine was placed at C-2 on the basis of its appearance at 3.52 ppm, as it is known that in an 8-ketospirobenzylisoquinoline a methoxy group at C-2 should appear at 3.50-3.65 whereas at C-3 appears at 3.70-3.94 ppm¹.

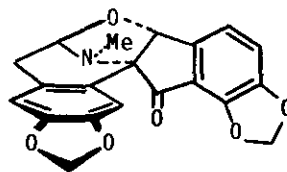
(+)-Africanine (3) was obtained as a white, crystalline, optically active substance, mp 237°C (MeOH), $[\alpha]_D^{25} +22^\circ$ (c= 0.322, CHCl₃). Its molecular formula



1 R=H
2 R=Me



3 R=Me
4 R=Ac



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Table 1. PMR and n.o.e. data for (+)-africanine 3 and (+)-africanine acetate 4

Proton	<u>3</u> (CDCl ₃ + TFA-d ₁)		<u>4</u> (CDCl ₃)	
	δ	n.o.e. (%) ^a	δ	n.o.e. (%) ^a
H-1	6.46(s)	H-13 (1.1)	6.62(s)	H-13 (1.7)
H-4	6.78(s)	OMe (6.7)	6.88(s)	OMe (9)
H-5	3.54(m)	H-6 (5.3); H-4 (2.1)		
H-6	4.32(m)	H-5 (2.9), N-Me (3.8)	2.9-3.7 (m)	(b)
H-11	7.25(d, J=7.8) [*]	(b)	7.14 (d, J=8)	(c)
H-12	7.23(d, J=7.8) [*]	(b)	7.13 (d, J=8)	(c)
H-13	6.05 (s)	H-1 (3.4)	5.39 (br s)	H-1(4), H-12(3.8)
N-Me	3.32 (s)	(b)	2.77 (s)	(c)
OMe	3.89 (s)	H-4 (3.7)	3.82 (s)	H-4 (3.6)
OCH ₂ O	6.28(d, J=1)	(b)	6.20 (d, J=1)	(b)
	6.26(d, J=1)	(b)	6.18 (d, J=1)	(b)
OAc	-----	-----	2.22 (s)	(b)

(a) Protons shown are the ones which experience enhancement upon pre-irradiation of the proton indicated in Column 1.

(b) non-irradiated signal.

(c) no n.o.e. was observed.

C₂₀H₁₇O₆N was established by high resolution MS (Found: 367.1044 \pm 0.0018; Calcd.: 367.1056) and fragments were also observed at 335 (50%) and 321 (100%). An intense absorption band in its IR spectrum at ν_{\max} (KBr) 1710 cm⁻¹, which is typical of a conjugated carbonyl in a five-member ring, its UV spectrum, with absorption maxima at λ_{\max} (EtOH) (log ϵ) 206 (4.29), 233 (4.42), 259 (sh, 4.00), 290 (3.62) and 346 (3.26) revealed the presence of an 8-ketospirobenzylisoquinoline skeleton^{1,4} in its structure. Upon addition of base its UV spectrum suffered a bathochromic shift to λ_{\max} (EtOH+OH⁻) 310 and 350 nm, indicating the phenolic nature of the alkaloid. Further support was obtained from its treatment with Ac₂O/Py, which afforded a simple compound. Its PMR (250 MHz, CDCl₃, δ) (2.22 ppm, s,

3H, -CO-CH₃) showed it to be monoacetylated.

The PMR spectrum (250 MHz, CDCl₃, δ) of (+)-africanine (3) exhibited signals at 2.75 (s, 3H, NMe), 3.20-3.60 (broad signals, 3H), 3.88 (s, 3H, OMe), 5.32 (s, 1H, HCO), 5.54 (br s, 1H, OH), 6.18 and 6.21 (AB dd, J= 1.0 Hz, OCH₂O) and 6.48 (s, 1H, H-1), 6.77 (s, 1H, H-4) and 7.14 (s, 2H, H-11 and H-12). All the above spectroscopic data were very similar to those of densiflorine (5)⁶, the only significant difference being the presence in (+)-africanine of a singlet at 3.88 (OMe) and 5.54 (OH) instead of the singlet at 5.93 ppm (OCH₂O) in densiflorine (5)⁶. Hence the only structural difference between the two alkaloids is the presence in (+)-africanine of methoxy and hydroxy groups in ring A instead of the methylenedioxy group of densiflorine. The position of the methoxy singlet (3.79 ppm) places the methoxy group at C-3 (see argument for (+)-isoparfumine (1) above).

Upon addition of one drop of deuterated trifluoroacetic acid (TFA-d₁) to a CDCl₃ solution of (+)-africanine, sharper signals for protons H-5 and H-6 were observed, along with a significant downfield shift of the H-6, H-13 and N-Me signals (see Table 1). Assignments of all the resonances were confirmed by nuclear Overhauser effect experiments (n.O.e.). In table 1 PMR data for (+)-africanine acetate are also given. Specially relevant is the n.O.e. observed for protons H-1 and H-12 after saturation of H-13, which are those expected for the spiro structure (4). Thus structure (3) is established for (+)-africanine, making it the second known sample of a spirobenzylisoquinoline alkaloid with an oxygen bridge between C-6 and C-13.

ACKNOWLEDGMENT

We thank the Comisión Asesora (Spain) for its financial support.

REFERENCES

1. R. M. Preisner and M. Shamma, J. Nat. Prod., 1980, 43, 305.
2. L. Castedo, A. Peralta, R. Suau and J. M. Saá, An. Quím., 1984, 80 (C), 264.
3. F. Šantavý, in "The Alkaloids" Vol. XVII, Academic Press, New York, 1979.
4. F. Šantavý, F. L. Hruban, V. Šimánek and D. Walterová, Collect. Czech. Chem. Commun., 1970, 35, 2418.
5. C. K. Yu and D. B. MacLean, Can. J. Chem., 1971, 49, 3025.

6. M. E. Popova, V. Simánek, J. Novák, L. Dolejs, P. Sedmera and V. Preininger,
Planta Medica, 1983, 48 (4), 272.

Received, 19th June, 1986